

## PRACTICE: CALCULATING pH OF ACIDS

1. Calculate the pH of a 0.0100 mol/L solution of hydrocyanic acid, HCN(aq)
2. Calculate the pH of a 0.12 mol/L solution of hydrobromic acid, HBr(aq)
3. Calculate the pH of a 0.25 mol/L solution of hydrofluoric acid, HF(aq)